Development of a Simple Reprocessing Process Using Selective Precipitant for Uranyl Ions

- Engineering studies on systems for precipitating and separating -

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Background Key Reaction: $UO_2^{2+} + 2NO_3^{-} + 2NCP$ $UO_2(NO_3)_2(NCP)_2$ FBR reprocessing with NCP



Mox fuel (U+Pu) & Blanket fuel (U) [DF100] for FBR reprocessing

Our approach for 3years(2002-2004)

Fundamental studies (beaker-scale)

· Precipitation mechanism

- ·Condition for precipitation
- ·Decontamination factors of FP...etc.
- ·Resistance to -ray radiation

Engineering studies

Scale-up experiment for

- precipitation
- precipitate separation
- Nuclear fuel manufacturing

Evaluation of the system



Outline



Fundamental studies

High recovery of uranium by precipitation
Decontamination factor (DF)

Engineering studies (2003-2004)

Design / construction of each apparatus and system

Engineering experiment

Objective of Apparatus Development

- Cost Reduction
 - · continuous operation
- Nuclear criticality safety
- Easy Maintenance ·long-term stability



Precipitator design Requirements

Nuclear criticality safety
 Continuous operation
 Long-term stability
 Sufficient mixing of reactants
 Avoiding NCP short-pass

Apparatus design

Overflow Tube NCP /Uranyl nitrate solution slurry Draft tube <u>Impeller</u> slurry

Point of the experiment

- Completely mixed Flow (reaction & withdrawal)
 High reactivity
 - (reaction rate, molar ratio)

 Column with tall-shape for criticality safe
 by shape component
 Draft tube installation with effective impeller design

Type : cylindrical-longitudinal Size: 100mm × L900mm Effective volume : 4L Feed rate :8L/hr 1/20-scale of 200tHM/y

Precipitate separator design

Requirements	Point of the experiment
Nuclear criticality safety Continuous operation	 High DF (target value, 100) Precipitate with low water
Long-term stability	content
 Rinsing (washing) ability for DF Avoiding leakage of solid to soln. 	Low leakage of precipitant to solution

Apparatus design



Equipment for the engineering studies



Precipitation experiment (1)



Precipitation experiment (2)

Results:

1. Stable withdrawals of slurry (mixed flow of precipitate and solution) were maintained during operation time (4 times longer than the stipulated residence time).



2. Sufficient mixing of reactants was achieved.

>95% of NCP was used for formation of UO_2^{2+} -NCP precipitate in 1st stage. >Over 99% of U was recovered as UO_2^{2+} -NCP precipitate in 2nd stage.



Precipitate separation experiment (1)



*Volume ratio of [rinse solution] / [precipitate]

Precipitate separation experiment (2) <u>Results</u>:

1.high DF ,Low water content



The rinsing operation is effective for increasing decontamination of FP.



In progress for over DF100 at the 1st stage

2.Low leakage of precipitant to solution Amount of precipitate in mother liquor was almost negligible.



Conclusion

- 1. Apparatus and system for precipitation and precipitate separation, which enable continuous operation and nuclear criticality safety, was designed.
- 2. In the precipitation experiment, sufficient mixing of reactants in precipitator and stable withdrawal of slurry were confirmed.
- 3. In the precipitate separation experiment, effectiveness of the rinsing operation in increasing DF of FP and low leakage of precipitant to mother liquor were confirmed.

Future remarks

- 1. Confirmation of long-term stability for equipment.
- 2. Adjusting the operational condition to achieve over 100 of DF in precipitate separation.



Image of NCP precipitate



a. Outlook



c. SEM image(x2000)



b. Microscope image



d. SEM image(x10000)

