



# **Research and Development for Innovative Partitioning System in COE**

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## **Background**

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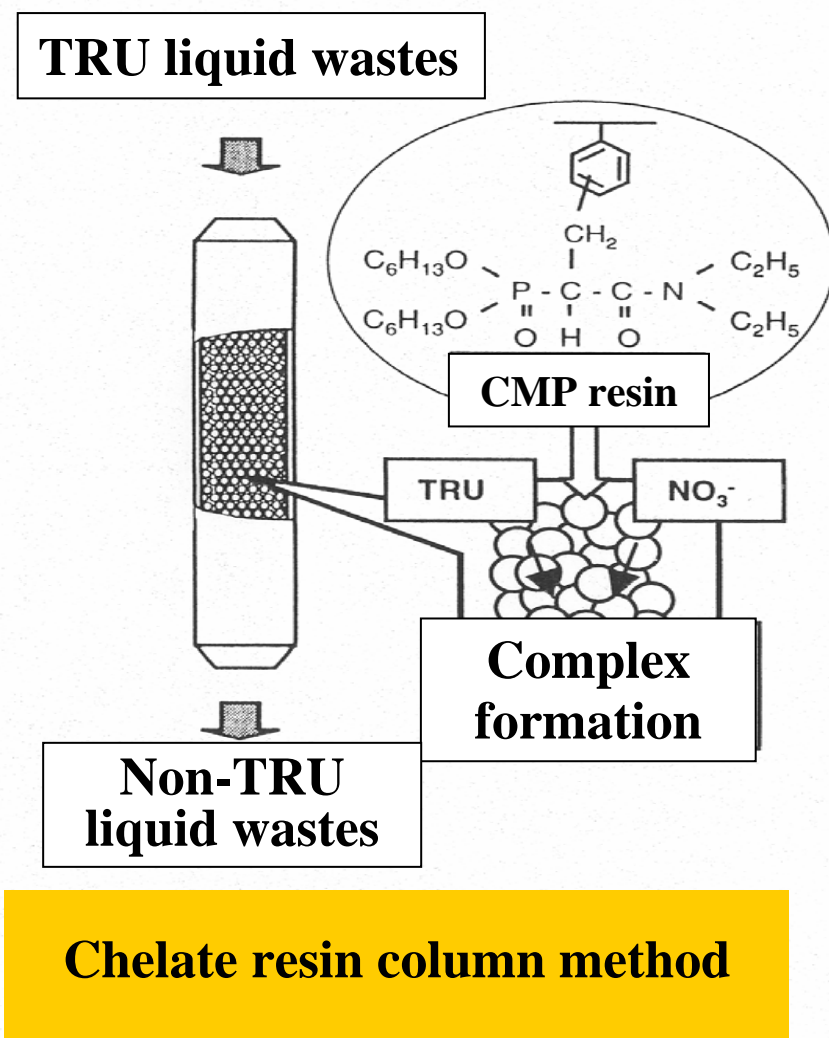
**In the concept of Innovative Nuclear Energy Systems for Sustainable Development of the World(COE-INES), we aim at the minimum-release of radioactive wastes from the nuclear energy park.**

**In order to achieve our aim, we need to do the following matters;**

- rational classification of nuclides**
- development of techniques for their partitioning**
- development of appropriate methods for management and disposal of radioactive wastes containing such nuclides.**



# Objective



**Usual partitioning methods:**  
Macroscopic separation methods by controlling phenomena in the bulk phases

- Liquid-liquid extraction
- Extraction chromatographic method
- Chelate resin method

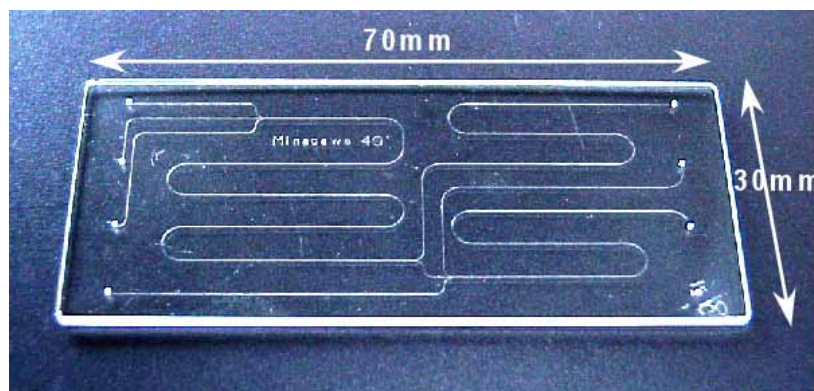
**Innovative methods using phenomena in micro- and nano-levels.**

To investigate the innovative partitioning system consisting of

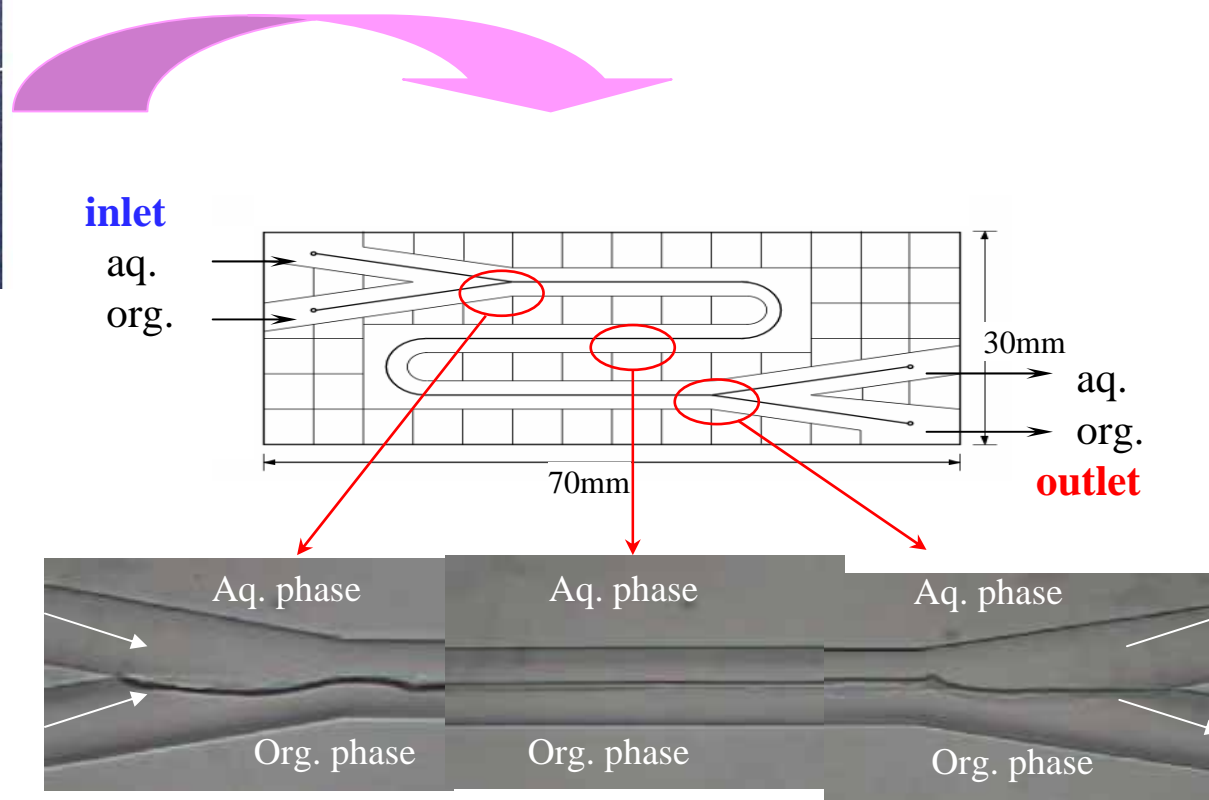
- the nuclide separation using microchannel chips
- the laser isotope separation by near-field optics
- high sensitive nano-level analytical methods



# Extraction separation of nuclides using microchannel chip



**Typical microchannel chip**  
(Channel size: 1- 200  $\mu\text{m}$ )

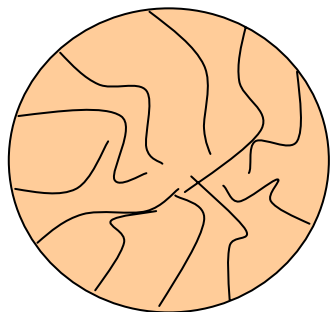


**Stable laminar flow**

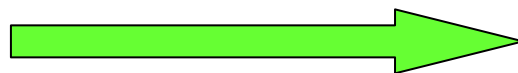




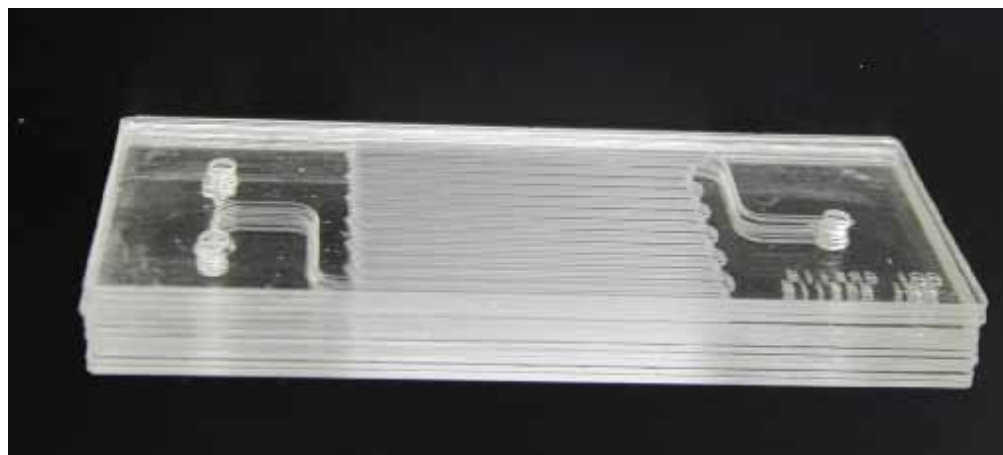
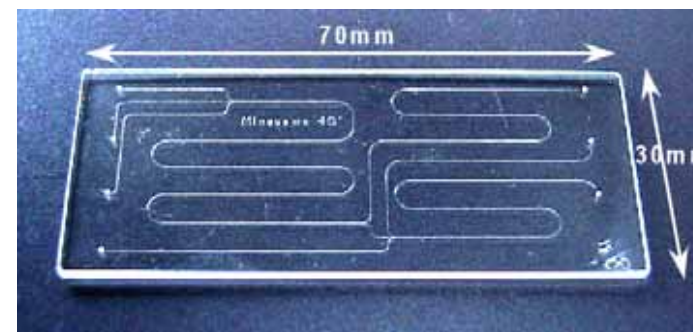
## Multilayer microchannel chip



**Ion exchange or  
chelate resin**



**Chip with microchannel  
similar to pores in ion  
exchange and chelate resins**

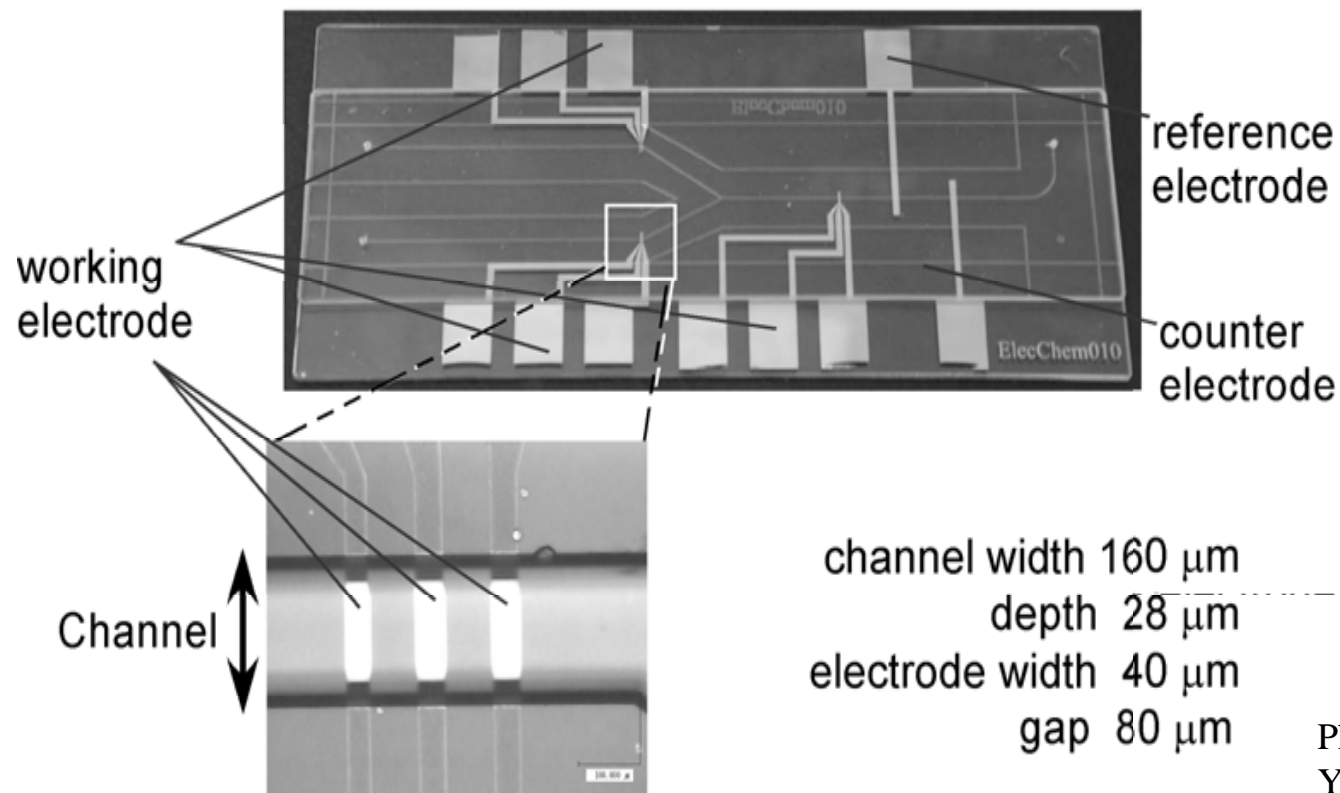


**In each chip, the selective adsorbents for  
specific nuclide are self-assembled onto the  
surfaces of microchannels.**

Photograph: provided by  
Dr. M. Tokeshi, Prof. T.  
Kitamori



## Valence adjustment using microchannel chip equipped with thin-layer electrodes



**The valences of nuclides can be adjusted by the electrodes in microchannels. This should lead to fine separation of nuclides.**

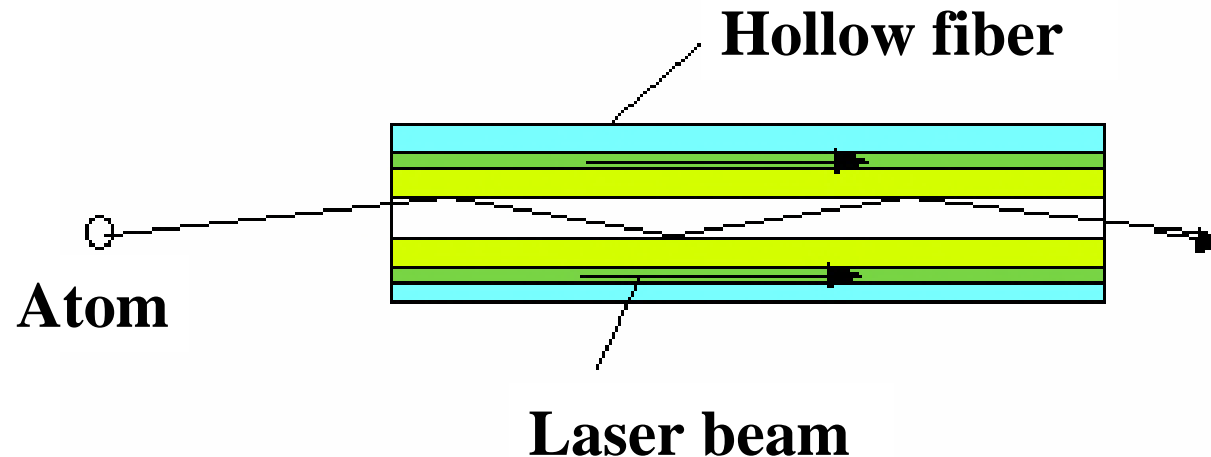




# Laser isotope separation by near-field optics

(by Prof. Akatsuka)

**This method should be applicable to the nuclides, which need isotope separation for transmutation.**



- Atoms are passed through hollow fiber.
- By selecting laser beam with appropriate wavelength, the dipole force induced by the near-field optics acts on the specific isotope as the repulsive force. Other isotopes are adsorbed on the inner surface of fiber.
- Inner diameter:  $\ll$  a few  $\mu\text{m}$ , length: a few cm, wavelength of laser: visible.

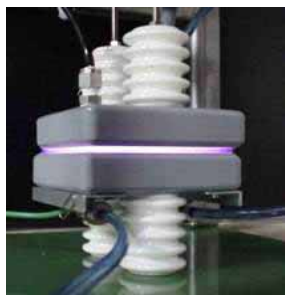




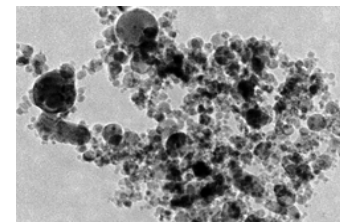
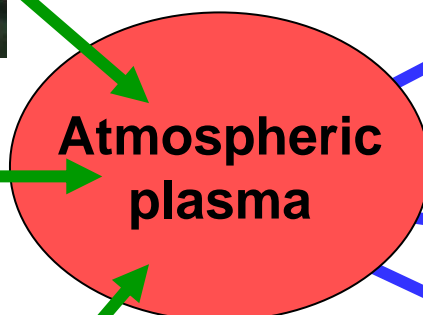
# R & D of nanomaterials for microchannel processes

(by Prof. Watanabe)

Atmospheric  
glow discharge



High frequency  
thermal plasma



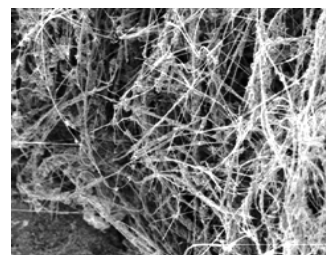
LaB6 nanoparticle

Selective separation

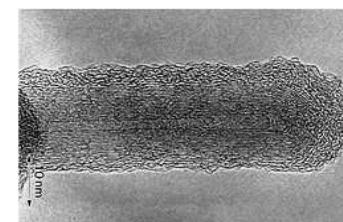
Material for microchannel

High speed nanoscale etching

High performance nanomaterial  
syntheses



Carbon nanocubes

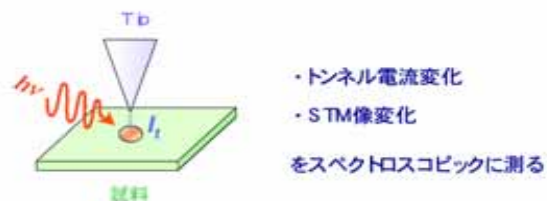


BCN nanochubes



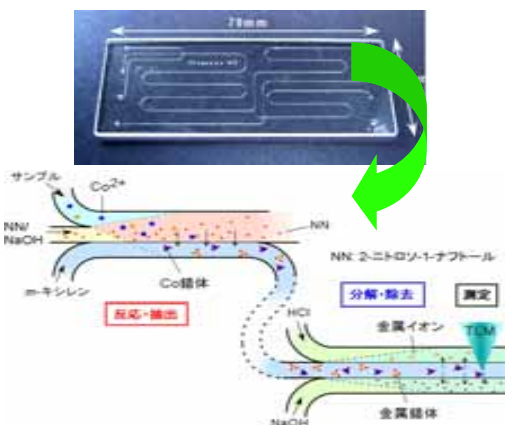


# High sensitive nano-level analytical methods (1): Development of a new apparatus of nano-spectromicroscopy for single-molecule science (by Prof. Onoe)



•Photoconductivity  
•Photovoltaic Effect  
•Charge State Change  
•Photo-chemical Reaction  
•Diffusion  
•Desorption  
etc.

Spatial resolution : 1-2 nm  
Energy resolution : 10 meV



Application to in situ  
observation in microchannel  
processes

Goal -> Top level of spatial and energy  
resolution

In situ single molecule observation of a  
metal complex under solvent free  
conditions (UHV)

DFT  
calculations

Effect of solvents on  
the electronic structure  
of the metal complex

To reveal a single-molecule  
behavior

Molecular-level understanding  
of extraction processes using  
nano-spectromicroscopy  
technique

Quantum design of ligand  
molecules for an  
innovative separation  
process

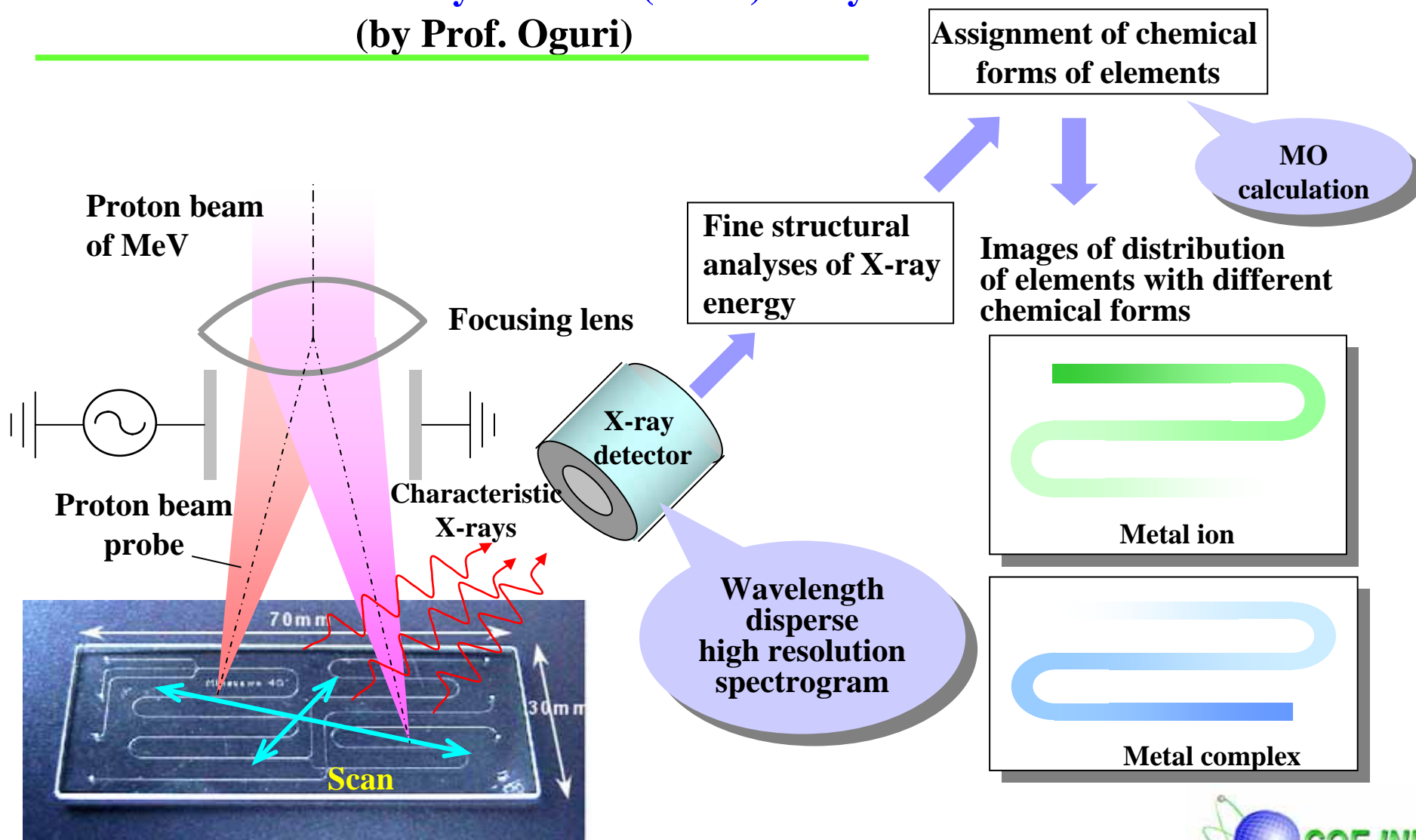




## High sensitive nano-level analytical methods (2):

### Particle-induced X-ray emission (PIXE) analysis

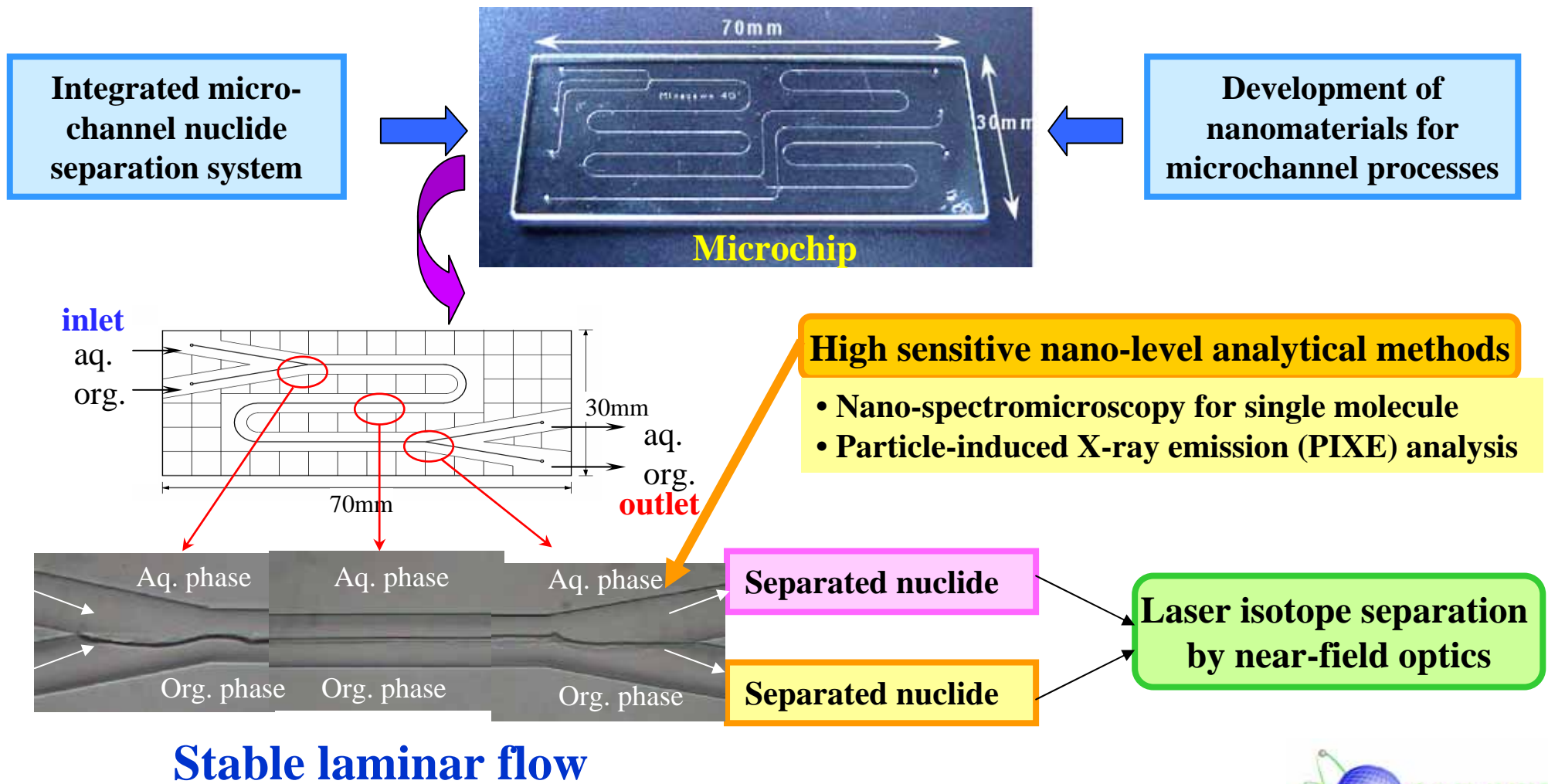
(by Prof. Oguri)







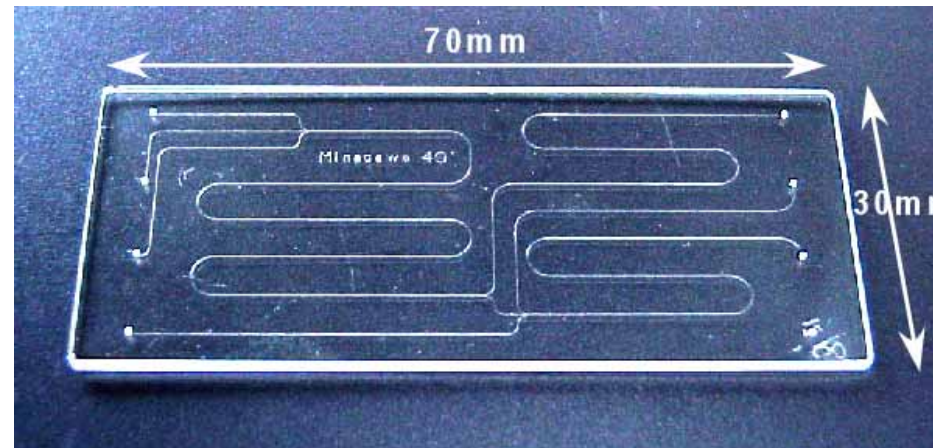
# Concept of innovative partitioning system based on micro- and nano-technologies







## Application of microchannel chip



Photograph: provided by  
Dr. M. Tokeshi, Prof. T.  
Kitamori

### Application of reactions in the space of 1 – 200 $\mu\text{m}$

#### 1. Narrow space

Short diffusion distance  $\rightarrow$  Short reaction time

#### 2. Large specific surface area

Large ratio of liquid-liquid contact area to  
feed reactants  $\rightarrow$  High efficient mixing and  
separation

#### 3. Small heat capacity

Easy control of temperature

### Comparison with conventional reactors (e.g.: comparison between cm and $\mu\text{m}$ scales)

**Scale: 1/100**

**Diffusion time: 1/10000**

**Reaction time: second order**



- Miniaturization of equipments and facilities
- Rapid operation